

## **Protein Structures**

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## Intrinsically Disordered p53 and Its Complexes Populate Compact Conformations in the Gas Phase\*\*



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The use of ion mobility mass spectrometry (IM-MS) to study the topology of proteins and their complexes is attracting considerable attention.<sup>[1]</sup> Under the appropriate conditions results have shown that unique insight can be gained for intractable systems such as protein aggregates, [2] viral capsids,[3] and membrane complexes.[4] From fundamental studies of peptides and proteins it is clear that folded structure, present in bulk solution, upon transfer into the gas phase, can retain native-like compact states but may unfold when activated or stored.[1,5] Similarly assemblies of globular proteins studied by IM-MS have been found to retain defined topological arrangements, [6] with unfolding of subunits being linked to activation in the gas phase. High charge states of proteins and their complexes in the gas phase are associated with extended unfolded structures, analogous to unfolding processes induced by protonation in solution.<sup>[1,7]</sup>

While the link between charge state and unfolding in the gas phase appears to hold for individual proteins and globular complexes it is not clear how the relationship between charge and structure would hold for complexes containing intrinsically disordered domains. Given that such domains are prevalent in nature (ca. 30% of the eukaryotic genome)[8] and have a high proportion of charged residues and a low percentage of hydrophobic ones, [9] these regions could remain elongated in the gas phase, analogous to the extended conformations of highly charged individual proteins. Alternatively in the absence of solvent these unstructured regions could collapse onto structured motifs. To examine these possible scenarios we applied IM-MS to one of the most studied natively disordered tetrameric complexes, that of the tumor suppressor protein p53. This protein complex, discovered more than 30 years ago, is a major chemotherapeutic

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target, [10] and has continued to thwart structural biologists largely because disordered regions comprise ca. 40% of the protein. [11] For our study we systematically introduced intrinsically disordered regions (flexible linker, N and C termini) to the two folded domains (Figure 1). This enabled us to probe the effects of disordered regions on the various constructs as well as full-length tetrameric p53 bound to cognate DNA.

The basic concept of IM-MS starts with injection of an ion packet into a cell filled with inert neutral gas. Aided by a weak electric field, the ions travel through the cell, colliding with neutral gas molecules in their path. Ions with a compact globular shape undergo fewer collisions and thus possess higher mobility than ions with more extended conformations, enabling their separation. Drift times measured in IM are usually converted into collision cross sections (CCS)—values that are independent of specific instrument parameters.<sup>[12]</sup> Theoretical CCS of candidate structures can be calculated either by a projection approximation (PA)<sup>[13]</sup> or an exact hard sphere scattering (EHSS)<sup>[14]</sup> algorithm. When high-resolution X-ray or NMR structures are available, a comparison between experimental and theoretical CCS can reveal different conformational states.<sup>[15]</sup> For many large protein assemblies, structural information is often only available for structured motifs, subunits or subcomplexes as is the case for p53.

In the absence of DNA, p53 is thought to exist as an ensemble of homo-tetramers with different coexisting conformations. Weakly associated dimers are formed by the core domain (C), which binds to DNA, and the tetramerization domain (T), which controls oligomerization. Upon DNA binding, p53 wraps around the double helix and a compact, rigid structure has been proposed. [11b,16] Starting with the simplest 4.1 kDa tetramerization domain (T) we applied IM-MS to the intact (T)<sub>4</sub> tetramers, the most abundant species observed from an ammonium acetate (AA) buffered solution (Figure 1 a and Figure S2). The average CCS obtained was in close agreement with theoretical values (Table 1). We next considered the isolated core domain (C), which was found to exist exclusively as 26.2 kDa monomers (Figure 1b), in accord with previous studies that show that C does not form higherorder oligomers in the absence of DNA.<sup>[17]</sup> Comparison of the experimental CCS with the theoretical value (Table 1) indicates a slightly collapsed native-like conformation of the p53 core, in line with previous IM data for this construct.<sup>[18]</sup> MS of the core-DNA complex (C)<sub>4</sub>DNA shows three species populated simultaneously: C, (C)<sub>2</sub>DNA, and (C)<sub>4</sub>DNA (Figure 1c). Monomeric C does not bind to DNA, and (C)<sub>2</sub>DNA and C are likely present in solution and not formed by collision-induced dissociation (CID) due to their charge state



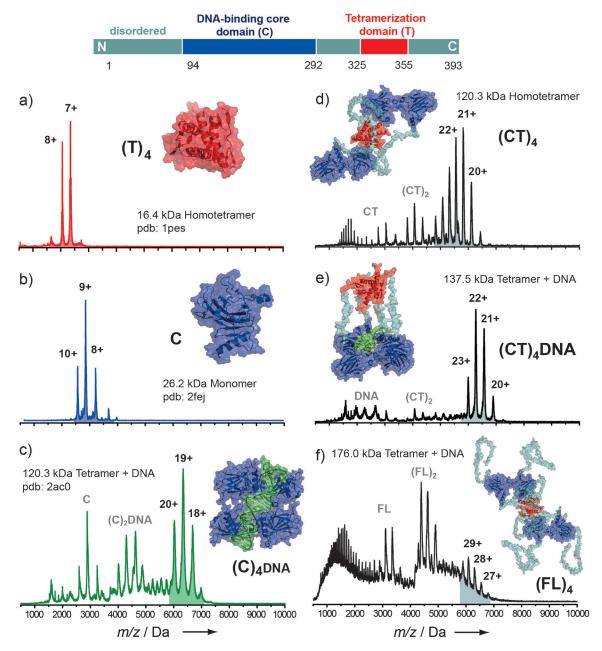


Figure 1. Schematic representation of p53 and mass spectra of individual domains and constructs. All spectra were recorded from ammonium acetate buffered solution and at similar instrument conditions. Structural models for  $(T)_4$ , C and  $(C)_4$ DNA correspond to X-ray or NMR structures with the pdb codes (a–c). Model structures for  $(CT)_4$ ,  $(CT)_4$ DNA, and FL<sub>4</sub> were obtained from a previous study. [11b]

signatures.<sup>[19]</sup> These observations are consistent with each dimer binding separately to a DNA repeat element in the dimer of dimers which constitute the p53 core.<sup>[20]</sup> Close agreement was found between theoretical and experimental CCS values for this (C)<sub>4</sub>DNA complex (Table 1).

Having obtained close correlation between theoretical and experimental CCS of (T)<sub>4</sub>, C and (C)<sub>4</sub>DNA, we considered the effects of the 14-amino acid disordered linker between C and T domains in the core-linker-tet tetramer (CT)<sub>4</sub> construct. The 120 kDa tetrameric, non-DNA containing form of the complex is accompanied by a small fraction of dimeric and monomeric CT (Figure 1 d). [8b, 14b] Surprisingly

the experimental CCS of (CT)<sub>4</sub> corresponds to ca. 60% of the theoretical values (Table 1). Interestingly there is little difference in the experimental CCS between (CT)<sub>4</sub> and (C)<sub>4</sub>DNA. This indicates a significant structural collapse of (CT)<sub>4</sub> in the gas phase, giving rise to a value close to the compact (C)<sub>4</sub>DNA structure. Adding DNA to (CT)<sub>4</sub> to form (CT)<sub>4</sub>DNA results in an increase in stability, as evidenced by the reduction in the population (CT)<sub>2</sub> (Figure 1e). The average CCS is much lower than theoretical values calculated for the proposed model structure of (CT)<sub>4</sub>DNA with extended disordered regions (Table 1). Comparing the experimental CCS of (CT)<sub>4</sub> and (CT)<sub>4</sub>DNA we find them to be closely

Table 1: Experimental and theoretical CCS for p53 constructs and their DNA complexes.[a]

| Complex           | Residues          | MW<br>[Da] | pdb   | Z  | CCS exp | CCS calcd [Å <sup>2</sup> ] |       |
|-------------------|-------------------|------------|-------|----|---------|-----------------------------|-------|
|                   |                   |            |       |    | [Ų]     | PA                          | EHSS  |
| (T) <sub>4</sub>  | 325-355           | 16400*     | 1pes  | 8  | 1453    | 1457                        | 1862  |
|                   |                   |            |       | 7  | 1397    |                             |       |
| С                 | 94–312            | 26150*     | 2 fej | 10 | 1746    | 1944                        | 2463  |
|                   |                   |            |       | 9  | 1851    |                             |       |
|                   |                   |            |       | 8  | 1907    |                             |       |
| (C)₄DNA           | 94–312<br>+ DNA** | 120100*    | 2ac0  | 20 | 5327    | 5397                        | 7098  |
|                   |                   |            |       | 19 | 5392    |                             |       |
|                   |                   |            |       | 18 | 5434    |                             |       |
| (CT) <sub>4</sub> | 94–360            | 120300     | ref.  | 22 | 5449    | 8184                        | 10319 |
|                   |                   |            | [116] | 21 | 5536    |                             |       |
|                   |                   |            |       | 20 | 5554    |                             |       |
|                   |                   |            |       | 19 | 5521    |                             |       |
| (CT)₄DNA          | 94–360            | 137500     | ref.  | 23 | 6012    | 8022                        | 10352 |
|                   | + DNA**           |            | [116] | 22 | 5978    |                             |       |
|                   |                   |            |       | 21 | 6092    |                             |       |
|                   |                   |            |       | 20 | 6105    |                             |       |
| (FL) <sub>4</sub> | 1–393             | 176 000    | ref.  | 29 | 7160    | 15 942                      | 19288 |
|                   |                   |            | [11b] | 28 | 7145    |                             |       |
|                   |                   |            |       | 27 | 7139    |                             |       |
|                   |                   |            |       | 26 | 7107    |                             |       |

[a] \*  $^{13}\text{C}/^{15}\text{N}$ -labeled variant, \*\* MW<sub>DNA</sub> 15 946 Da; for details see Supporting Information.

similar. Therefore it is likely that the structural collapse observed in both complexes is not due to the subdomains T<sub>4</sub> and C<sub>4</sub>, but rather to the connecting disordered linker.

We next considered the effects on the CCS of adding further intrinsically disordered regions. Full-length tetrameric p53 complex contains extensive, intrinsically disordered domains at the N and C termini and, similar to (CT)4, forms tetramers also in the absence of DNA. In mass spectra three major species are populated: FL monomer with a bimodal charge state distribution, (FL)<sub>2</sub> and (FL)<sub>4</sub> (Figure 1 f). The coexistence of (FL)<sub>2</sub> and (FL)<sub>4</sub> is in accord with p53 being a weakly associated dimer of dimers. The presence of FL monomers with high charge states is indicative of unfolding implying either gas phase activation and/or solution dissociation, in agreement with the low thermal stability of fulllength p53.<sup>[17]</sup> The experimental CCS of (FL)<sub>4</sub> is ca. 50% lower than the theoretical value.

To investigate the origin of this drastic difference between experimental and theoretical CCS we performed molecular dynamics (MD) simulations in vacuo using models of (CT)<sub>4</sub> and the (CT)<sub>4</sub>DNA as starting points. [11b] Simulations, three replicates (1 ns), showed that both complexes equilibrate towards compact conformations after ca. 0.1 ns. Excellent agreement was found between experimental CCS and the average values calculated for the most populated conformational states of (CT)<sub>4</sub> and (CT)<sub>4</sub>DNA during MD (Table 1, Figure 2). The backbone root mean square deviation (RMSD) from corresponding starting structures was calculated over the simulation period. The RMSD of the individual

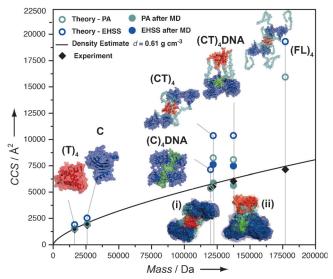


Figure 2. Plot of mass against CCS for p53 domains and constructs together with representative structures from MD simulations. Experimental values (black diamonds) are compared with theoretical CCS obtained by the projection approximation (PA, light blue open circles) and exact hard sphere scattering (dark blue open circles, EHSS) algorithms (Table 1) using the MOBCAL software. [13b] CCSs are calculated by using atomic information from pdb files (Table 1) or proposed models.[116] Analogous values are calculated after MD simulation (filled circles). The estimated mass-to-CCS correlation of idealized spherical particles, with the average density of a native-like gas-phase protein, is shown (line). [12b] Three of the most populated conformers formed during MD simulations of (CT) $_4$  and (CT) $_4$ DNA are overlaid: (i) and (ii), respectively.

core and tetramerization domains within the (CT)<sub>4</sub> and (CT)<sub>4</sub>DNA complexes was found to increase by 3–4 Å. By contrast the full complexes exhibited a significant RMSD from their initial conformations (16-20 Å, Figure S3). This value is considerably larger than the 2-3 Å RMSD observed in silico for other proteins and complexes in solvent free environments.<sup>[21]</sup> This implies that rearrangement of p53 constructs arises from convergence of the essentially unaltered core and tetramerization domains, by collapse of the flexible linkers.

In principle, all proteins or protein complexes might undergo a compaction when transferred into a solvent-free environment. This collapse is characterized by minor rearrangements of the charged residues at the protein exterior, which fold back to the protein surface to form strong electrostatic interactions upon removal of water. [22] For large protein complexes such a compaction would produce only a subtle effect. More pronounced reorganizations have been reported in which complexes undergo spontaneous compaction when transferred into the gas phase [3,23,25] or collapse in response to collisional activation. [15,21a,23,24,26] In both scenarios, the extent of collapse is related to the topology of the complex and the major difference between the two was shown to be the charge state dependence. [21a] For the p53 constructs used here, no correlation between charge state and collapse was observed (see also the Supporting Information) implying that compaction occurs spontaneously.

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The average gas-phase density of native-like, globular proteins is ca. 0.61 g cm<sup>-3</sup> enabling the CCS of an idealized spherical protein to be calculated.<sup>[12]</sup> Comparing CCSs for p53 constructs as well as the constant-density estimate (Figure 2) reveals a clear correlation between measured values and CCSs estimated for spherical particles. Thus despite extensive collapse, p53 constructs exhibit typical packing densities of native-like globular proteins. Collapsed native-like structure is further indicated by in vacuo MD simulations of (CT)<sub>4</sub> and (CT)<sub>4</sub>DNA (Figure 2). CCSs calculated for the compact MDderived structures agree with values obtained experimentally. Interestingly, backbone RMSD analysis, during collapse, is consistent with the majority of residues in the folded C and T domains remaining unaffected. By contrast considerable rearrangements are observed for residues in the intrinsically disordered linker (Figure S3).

Given this drastic collapse it is interesting to consider at what stage this occurs. If we examine the average charge state of all complexes we find that (T)4, C, and (C)4DNA show good agreement to predictions based on the correlation with surface area<sup>[27]</sup> but (CT)<sub>4</sub> and (CT)<sub>4</sub>DNA reveal significantly lower charge states than predicted (Table S2). This is in contrast to isolated, monomeric intrinsically disordered proteins, for which the charge states were shown to correlate well with surface area estimates.<sup>[28]</sup> It is noteworthy that the surface areas calculated for the collapsed structures, formed during MD simulations, recapitulate the experimental average charge state remarkably closely (Table S2) providing further evidence for collapse. Interestingly, loss of surface area of the native-like p53 constructs implies that collapse occurs concomitantly with charging. This is in accord with recent proposals for the electrospray mechanism in which complexes reside in the interior of droplet and solvent and electrolyte ions expel the excess surface charge of droplets. Charging of the protein takes place at the very last stages of desolvation, by transfer of charges onto the gas phase complex. [29] Consequently, fewer charges than expected from the surface area estimate are obtained for the p53 constructs with large intrinsically disordered regions, which must therefore collapse very late in the electrospray process.

The magnitude of the collapse observed here is unprecedented among protein complexes considered to date. This may be attributed in part to overestimation of the intrinsically disordered domains in the CCS calculations. However such an effect is unlikely to account for the differences in CCS and average charge of the magnitude observed here. We conclude that the vast majority of the gas phase rearrangement of p53 occurs at the flexible linker and the disordered termini, which exhibit similar properties to unfolded protein chains that collapse in the gas phase. Thus the CT and FL p53 complexes do not retain the same conformational state in the gas phase as observed in solution. The inner core, by contrast, which governs the intermolecular organization of the complex, appears to retain its structural integrity.

## **Experimental Section**

Protein purification and expression is described in the Supporting Information. All proteins and complexes were buffer-exchanged into

500 mm AA, pH 6.9 except FL (250 mm AA, pH 6.9) using Micro Bio-Spin 6 columns (Bio-Rad, UK) or Slide-A-Lyzer dialysis cassettes (Fisher Scientific, UK). As charge reduction can stabilize complexes without noticeably affecting their folding behavior in solution, AA buffer was in some cases replaced with triethylammonium acetate (TEAA) buffer. [30] In these conditions lower charged ions were formed (Figure S2). The CCSs remained similar (Table S1). DNA, added after buffer exchange, is the p53 response element. [11b] Absolute CCS measurements were performed on a modified Synapt HDMS<sup>[31]</sup> described in detail previously. [12b]

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- B. C. Bohrer, S. I. Merenbloom, S. L. Koeniger, A. E. Hilderbrand, D. E. Clemmer, Annu. Rev. Anal. Chem. 2008, 1, 293

  327.
- [2] S. L. Bernstein, N. F. Dupuis, N. D. Lazo, T. Wyttenbach, M. M. Condron, G. Bitan, D. B. Teplow, J. E. Shea, B. T. Ruotolo, C. V. Robinson, M. T. Bowers, *Nat. Chem.* 2009, 1, 326–331.
- [3] C. Uetrecht, I. M. Barbu, G. K. Shoemaker, E. van Duijn, A. J. R. Heck, *Nat. Chem.* 2011, 3, 126–132.
- [4] M. Zhou, N. Morgner, N. P. Barrera, A. Politis, S. C. Isaacson, D. Matak-Vinkovic, T. Murata, R. A. Bernal, D. Stock, C. V. Robinson, *Science* 2011, 334, 380–385.
- [5] L. W. Zilch, D. T. Kaleta, M. Kohtani, R. Krishnan, M. F. Jarrold, J. Am. Soc. Mass Spectrom. 2007, 18, 1239 – 1248.
- [6] B. T. Ruotolo, K. Giles, I. Campuzano, A. M. Sandercock, R. H. Bateman, C. V. Robinson, *Science* 2005, 310, 1658–1661.
- [7] H. Shi, N. A. Pierson, S. J. Valentine, D. E. Clemmer, J. Phys. Chem. B 2012, 116, 3344-3352.
- [8] R. Pancsa, P. Tompa, PLoS One 2012, 7, e34687.
- [9] V. N. Uversky, Int. J. Biochem. Cell Biol. 2011, 43, 1090-1103.
- [10] A. J. Levine, M. Oren, Nat. Rev. Cancer 2009, 9, 749-758.
- [11] a) E. N. Nikolova, E. Kim, A. A. Wise, P. J. O'Brien, I. Andricioaei, H. M. Al-Hashimi, *Nature* 2011, 470, 498–502; b) H. Tidow, R. Melero, E. Mylonas, S. M. V. Freund, J. G. Grossmann, J. M. Carazo, D. I. Svergun, M. Valle, A. R. Fersht, *Proc. Natl. Acad. Sci. USA* 2007, 104, 12324–12329.
- [12] a) B. T. Ruotolo, J. L. P. Benesch, A. M. Sandercock, S.-J. Hyung, C. V. Robinson, *Nat. Protocols* 2008, 3, 1139–1152;
  b) M. F. Bush, Z. Hall, K. Giles, J. Hoyes, C. V. Robinson, B. T. Ruotolo, *Anal. Chem.* 2010, 82, 9557–9565.
- [13] a) G. von Helden, M. T. Hsu, N. Gotts, M. T. Bowers, J. Phys. Chem. 1993, 97, 8182-8192; b) M. F. Mesleh, J. M. Hunter, A. A. Shvartsburg, G. C. Schatz, M. F. Jarrold, J. Phys. Chem. 1996, 100, 16082-16086.
- [14] A. A. Shvartsburg, M. F. Jarrold, Chem. Phys. Lett. 1996, 261,
- [15] B. T. Ruotolo, S.-J. Hyung, P. M. Robinson, K. Giles, R. H. Bateman, C. V. Robinson, *Angew. Chem.* 2007, 119, 8147–8150.
- [16] E. Natan, D. Hirschberg, N. Morgner, C. V. Robinson, A. R. Fersht, *Proc. Natl. Acad. Sci. USA* 2009, 106, 14327 – 14332.
- [17] A. C. Joerger, A. R. Fersht, Annu. Rev. Biochem. 2008, 77, 557 582.
- [18] P. A. Faull, H. V. Florance, C. Q. Schmidt, N. Tomczyk, P. N. Barlow, T. R. Hupp, P. V. Nikolova, P. E. Barran, *Int. J. Mass Spectrom.* 2010, 298, 99–110.
- [19] J. L. P. Benesch, J. Am. Soc. Mass Spectrom. 2009, 20, 341 348.
- [20] E. Arbely, E. Natan, T. Brandt, M. D. Allen, D. B. Veprintsev, C. V. Robinson, J. W. Chin, A. C. Joerger, A. R. Fersht, *Proc. Natl. Acad. Sci. USA* 2011, 108, 8251–8256.



- [21] a) Z. Hall, A. Politis, M. F. Bush, L. J. Smith, C. V. Robinson, J. Am. Chem. Soc. 2012, 134, 3429-3438; b) R. Friemann, D. S. D. Larsson, Y. Wang, D. van der Spoel, J. Am. Chem. Soc. 2009, 131, 16606 - 16607.
- [22] K. Breuker, F. W. McLafferty, Proc. Natl. Acad. Sci. USA 2008, 105, 18145 - 18152.
- [23] C. J. Hogan, B. T. Ruotolo, C. V. Robinson, J. Fernandez de La Mora, J. Phys. Chem. B 2011, 115, 3614-3621.
- [24] T. W. Knapman, V. L. Morton, N. J. Stonehouse, P. G. Stockley, A. E. Ashcroft, Rapid Commun. Mass Spectrom. 2010, 24, 3033 -
- [25] C. Kaddis, S. Lomeli, S. Yin, B. Berhane, M. Apostol, V. Kickhoefer, L. Rome, J. Loo, J. Am. Soc. Mass Spectrom. 2007, 18, 1206-1216.

- [26] J. Freeke, C. V. Robinson, B. T. Ruotolo, Int. J. Mass Spectrom. **2010**, 298, 91-98.
- [27] Z. Hall, C. V. Robinson, J. Am. Soc. Mass Spectrom. 2012, 23, 1161 - 1168.
- [28] L. Testa, S. Brocca, R. Grandori, Anal. Chem. 2011, 83, 6459-6463.
- [29] C. J. Hogan, Jr., J. A. Carroll, H. W. Rohrs, P. Biswas, M. L. Gross, Anal. Chem. 2009, 81, 369-377.
- [30] a) D. Lemaire, G. Marie, L. Serani, O. Laprevote, Anal. Chem. 2001, 73, 1699-1706; b) K. Pagel, S.-J. Hyung, B. T. Ruotolo, C. V. Robinson, Anal. Chem. 2010, 82, 5363-5372.
- [31] S. D. Pringle, K. Giles, J. L. Wildgoose, J. P. Williams, S. E. Slade, K. Thalassinos, R. H. Bateman, M. T. Bowers, J. H. Scrivens, Int. J. Mass Spectrom. 2007, 261, 1-12.

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